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Potentiometric And Spectrophotometric Determination Of

A spectrophotometric and a potentiometric FIA method were used to determine calcium in samples of back water from paper mills. The spectrophotometric method used the complexation reaction between calcium and o-cresolphthalein complexon. Optimum pH for the method was calculated theoretically.

Potentiometric and spectrophotometric determination of ...

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Potentiometric and spectrophotometric determination of ...

The protonation constant of hexamethylenetetramine (urotropine) was determined by a potentiometric and a spectrophotometric method. The calculations gave log K HL (concentration constants): 4.89 at $\mu = 0.1$ and 5.05 at $\mu = 0.5$. The temperature was 25° and potassium chloride was used to adjust the ionic strength.

Potentiometric and spectrophotometric determination of the ...

The MDM-water mixture is a new multicomponent cosolvent mixture (consisting of equal volumes of methanol, dioxane and acetonitrile, as organic solvents) that dissolves a wide range of poorly water-soluble compounds. The cosolvent dissociation constants ($p(s)K_a$) of 50 chemically diverse compounds (acids, bases and ampholytes) were measured in 15-56 wt% MDM-water mixtures by potentiometric or spectrophotometric titration and the aqueous pK_a values obtained by extrapolation.

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The Potentiometric and spectrophotometric determination of phosphoric acid content was conducted on three batches of selected beverages: Regular and Light Coke, Regular and Light Pepsi, Smirnoff Ice and 7 Up with 7 Up as control. The phosphoric acid.

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New, simple and convenient potentiometric and spectrophotometric methods are described for the determination of dextromethorphan hydrobromide (DXM) in pharmaceutical preparations. The potentiometric technique is based on developing a potentiometric sensor incorporating the dextromethorphan tetrakis(p-chlorophenyl)borate ion-pair complex as an ...

New potentiometric and spectrophotometric methods for the ...

The obtained results show the good reproducibility of potentiometric and spectrophotometric pK_a determination in MDM-water mixtures. The standard deviation values varied between 0.01 and 0.08 with an average value of 0.02 in absolute value. Table 2 indicates S.D. values in parenthesis as the least significant digit. Table 2.

Potentiometric and spectrophotometric pKa determination of ...

Conductometric, Potentiometric, and Spectrophotometric Determination of Dissociation Constants of Substituted Benzoic Acids in Acetonitrile 1 I. M. Kolthoff and

Conductometric, Potentiometric, and Spectrophotometric ...

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For the pK_a determination by the spectrophotometric method we measured the ratio between the concentration of dissociated and undissociated forms. The ratio was calculated from spectral data. The potentiometric method for pK_a consists of measuring the pH values within a potentiometric titration with 0.1 M NaOH.

The potentiometric and spectrophotometric determination of ...

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Potentiometric and spectrophotometric determination of the ...

DECFAM—A new computer oriented algorithm for the determination of equilibrium constants from potentiometric and/or spectrophotometric measurements—I. Computers & Chemistry 1984, 8 (2) , 91-99. DOI: 10.1016/0097-8485(84)85002-0.

Conductometric, Potentiometric, and Spectrophotometric ...

The potentiometric titration data were used to calculate the protonation constants of a secondary amino group in each drug, and the spectrophotometric titration data were used to determine the dissociation constants of OH groups influencing the distribution of electron density in the chromophore (aromatic system).

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Potentiometric and spectrophotometric pK a determination of water-insoluble compounds : Validation study in a new cosolvent system. / Völgyi, Gergely; Ruiz, Rebeca; Box, Karl; Comer, John; Bosch, Elisabeth; Takács-Novák, Krisztina. In: Analytica Chimica Acta, Vol. 583, No. 2, 05.02.2007, p. 418-428. Research output: Contribution to journal > Article

Potentiometric and spectrophotometric pKa determination of ...

A method has been developed for the quantitative determination of bithiosemicarbazones in which addition of copper(II) chloride to a solution of the organic compound leads to complex formation. The hydrogen ions liberated on complex formation are determined by potentiometric titration. The complex, or the organic