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SHORT ANSWER Answer the following questions in the space provided. 1. Match the description on the right to the most appropriate quantity on the left. d 2 m³ (a) mass of a small paper clip a 0.5 g (b) length of a small paper clip f 0.5 kg (c) length of a stretch limousine e 600 cm² (d) volume of a refrigerator compartment

2 Measurements and Calculations

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SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode . 2. The smallest unit of an element that can exist either alone or in molecules containing the same or different elements is the .atom 3.

3 Atoms: The Building Blocks of Matter

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev’s original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

5 The Periodic Law

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SHORT ANSWER Answer the following questions in the space provided. 1. For elements in their standard state, the value of H 0 f 0 is . 2. The formation and decomposition of water can be represented by the following thermochemical equations: H 2(g) 1 2O 2(g) → H 2O(g) 241.8 kJ/mol H 2O(l) 241.8 kJ/mol → H 2(g) 1 2O 2(g) taken in a.

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CHEMICAL FORMULAS AND CHEMICAL COMPOUNDS59

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