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SECTION 19.3 STRENGTHS OF ACIDS AND BASES(pages 605-611) This section defines strong acids and weak acids, and then explains how to calculate an acid dissociation constant. It describes how acids and bases are arranged by strength according to their dissociation constants (K a) and (K b). Strong and Weak Acids and Bases (pages 605-609) 1.

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Chapter 19 Acids, Bases, and Salts 501 5. Define the equivalence point in this reaction. What are [NaOH], [C₆H₅COOH], and [C₆H₅COONa] at the equivalence point? 6. What is the concentration of benzoic acid in the original “unknown” solution? 7. Refer to Figure 19.8 in your textbook. Which of the acid–base indicators shown

Name Date Class INTERPRETING GRAPHICS 19 Use with Section 19

•Myristic acid occurs in butterfat and in coconut oil. •Other fatty acids—called monounsaturated or polyunsaturated fatty acids—have one or more double bonds in their carbon chains. •Oleic acid is an example of a monounsaturated fatty acid; its formula is CH₃(CH₂)₇CH:CH(CH₂)₇COOH. •Oleic acid occurs in olive oil, peanut oil ...

Chapter 19 Biochemistry

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19.1 Acid-Base Theories> - Quia

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Omega-3 fatty acids (like those in Figure 3.17) fall into this category and are one of only two known for humans (the other is omega-6 fatty acid). These are polyunsaturated fatty acids and are omega-3 because a double bond connects the third carbon from the hydrocarbon chain's end to its neighboring carbon.

3.3 Lipids - Biology 2e | OpenStax

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