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For the atomic radii, it looks like a rollercoaster. There are 2 peaks, one at the top and one on the valley. The different peaks are on completely opposite sides of the periodic table. For the electronegativity graph, it shows how the different blocks on the periodic table act. The s and p blocks create peaks and the d blocks create humps.

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Unit 3 Review Sheet Key - FCPS

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Atomic Structure and Periodic Table Test Review

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Periodic Table of the Elements Notes Arrangement of the known elements based on atomic number and chemical and physical properties. Divided into three basic categories: Metals (left side of the table) Nonmetals (right side of the table) Metalloids (touching the zig zag line) Basic Organization by: Atomic structure

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