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82 ChemQuest 28 Name:

the same exponents. Your answer will them have the same exponent as the numbers you add or subtract. Here are some examples. Make sure you understand each step: 4.2x10 6 + 3.1x10 5 make exponents the same, either a 5 or 6 42x10 5 + 3.1x10 5 = 45.1x10 5 = 4.51x10 6

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Hint: add your answer to question 1 and four times your answer to question2. n\$ t ?(q\ = BZ e*eeAronS Verify that there are 32 electrons pictured in the drawing of CClq. 3es Find the sum of all the valence electrons for NF:. (Add how many valence electrons one nitrogen atom has with the valence electrons for three fluorine atoms.) H:?crl 2 Tte]e.

Critical Thinking Questions - LTHS Answers

PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 - HW Practice Keys - ChemActivity 51: Cell Voltage - ChemActivity 50: Electrochemical Cell UNIT 11 - HW Practice Keys - ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

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Cu(NO3)2 + Zn) Zn(NO3), + Cu There is a sentence that you can \$,rite to describe what happens above: Copper1D nitrate reacts withzinc metal and yields zinc nitrate and copper metal. Take a minute to notice the relationship between the sentence and the symbols in the reaction

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Hint: add your answer to question 1 and four times your answer to question 2. 4 + 4(7) = 4 + 28 = 32 4. Verify that there are 32 electrons pictured in the drawing of CCl4. Count each dot in the picture above: there are 24. There are also 8 more electrons represented by the 4 lines (each line

represents two electrons). 5.

ChemQuest 23 : Drawing Covalent Compounds

The ChemQuest tank farm of over 50 bulk storage units affords us the ability to receive, process, warehouse and transport various raw materials and solutions in a single, turn-key environment. Packaging & Labeling. Specialized chemical packaging and labeling is available through ChemQuest Chemicals.

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