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___ 5. Mendeleev is credited with developing the first successful. a. periodic table. b. method for determining atomic number. c. test for radioactivity. d. use of X rays. ___ 6. Mendeleev did not always list elements in his periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c ...

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Chemistry Chapter 5 Exam Multiple
Choice Identify the letter of the choice
that best completes the statement or
answers the question. ____ 1. The idea of
arranging the elements in the periodic
table according to their chemical and
physical properties is attributed to a.
Mendeleev. c. Bohr. b. Moseley. d.
Ramsay. ____ 2.

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Chapter Test Assessment Chapter Test A

Teacher Notes and Answers 5 The

Periodic Law TEST A 1. b 2. d 3. b 4. b 5.

d 6. a 7. b 8. b 9. b 10. a 11. c 12. a 13. c

14. d 15. c 16. b 17. d 18. a 19. d 20. c

21. d 22. a 23. a 24. d 25. b

Assessment Chapter Test A - Wag & Paws

This answer should then be converted to standard scientific notation. $32.15 \times 10^2 =$

$3.215 \times 10^3 = 3215$. 33. a. 124 (the

answer can only be given to three

significant figures because 0.995 is only

given to three significant figures) b.

1.995 $\times 10^{-23}$ (the answer can only be

given to four significant figures because

Chapter 5 Measurements and Calculations

Holt McDougal Modern Chemistry

Chapter Test Assessment Chapter Test B

Teacher Notes and Answers 5 The

Periodic Law TEST B 1. a 2. c 3. d 4. d 5.

a 6. a 7. c 8. a 9. lanthanides 10. 2 11.

fourth 12. transition elements 13. 32 14.

valence electrons 15. electron affinity

16. electronegativity 17. ionization

energy 18. $3s^2 3p^4$ 19. atomic radius 20

...

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5.1 Light and Quantized Energy MAIN Idea Light, a form of electromagnetic radiation, has characteristics of both a wave and a particle. 5.2 Quantum Theory and the Atom MAIN Idea Wavelike properties of electrons help relate atomic emission spectra, energy states of atoms, and atomic orbitals. 5.3 Electron Configuration MAIN Idea A set of three rules