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Atomic
number, mass
number and RAM
FORM 2 CHEMISTRY
THE STRUCTURE OF
AN ATOM \u0026amp;
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1. Study the flow chart below and answer the questions that follow:
4. Name reagent Z. 5. Describe the process which takes place in step 2. 6. Identify the white solid. 2. a) Starting from solid magnesium oxide, describe how a solid sample of magnesium hydroxide can be prepared. b) Give one

use of magnesium hydroxide. 3.

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Answers. Previous
Question Next
Question. Question .
Answers. Previous
Question Next
Question. Hello Guest,
talk to us. Online Close.
Hello Guest, talk to us.
Please name and try
again Message Sent!
We will get in touch
soon. If Urgent, call or
text 0716 858334 /
0712 054195.

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FORM 2 CHEMISTRY
TIME: 2 Hrs
INSTRUCTIONS
Attempt all the
question in the

question in the spaces provided All the working must be clearly shown where appropriate.

Negligence will be penalized. QUESTIONS

1.The table below shows some information concerning partials of certain elements. Element Atomic number Number of electrons Mass Number P 19 18 39

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1.1 The structure of the atom. When scientists started exploring matter, they realised that matter can be divided into smaller and still smaller particles. They called the smallest particle an 'atom'. The name 'atom' was derived from the Greek word 'atomos', meaning 'indivisible'.

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2013 INSTRUCTIONS TO CANDIDATES 1.

Answer all the questions in the spaces provided. 2.

Mathematical tables and electronic calculators may be used. 3. All working must be clearly shown where necessary 1. The table below shows the pH values of solutions J to N.

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1. What is the first element on the periodic table? 2. What is the centre of an atom called? 3. True or false? Acids have a pH level below 7. 4. What is the main gas found in the air we breathe?

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Boy (8-10) in Chemistry Lab. Charles Thatcher, Getty Images
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Dini ya Kiislamu.
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Biology Notes Form 2.
1. a) i) Define transport. Movement of substances from one part of the body to another. ii) Explain the necessity of transport in plants and animals. make nutrients move from one point to another. movement of respiratory gases i.e. oxygen and carbon IV oxide. elimination of metabolic wastes.