
Suicide Notes By Michael Thomas Ford

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Unit 6: Reactions and Stoichiometry

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Chapter 12 Stoichiometry Packet Answer Key Answer: 4.93×10^{-5} L or 49.3 μ L In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent: $[\text{Au}(\text{CN})_2]^-$, LaCl_3 , ethanol, and para -nitrophenol.

Stoichiometry Packet Answers**Chapter 12**

Chemical Reactions and Stoichiometry. Given the equation $3\text{A} + \text{B} \rightarrow \text{C} + \text{D}$, you react 1 mole of A with 3 moles of B. True or false: B; is the limiting reactant because you have fewer moles of B than A. Answer: False. NaHCO_3 ; is the active ingredient in baking soda. How many grams of oxygen are in 0.35 g of. NaHCO_3 ? a) 0.012 g. b) 0.022 g. c) 0.067 g. d) 0.20 g

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