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MCAT Question of the Day: Specific Heat Calculations

Calorimetry Examples: How to Find Heat and Specific Heat Capacity Calculations involving heat and specific heat How to calculate specific heat: Example specific heat problems Specific Heat Calculations—Heat 001 *Calculating Heat Combining Heat Calculations (Question 2)* Thermodynamics: Specific Heat Capacity Calculations **How to Calculate the Specific Heat Capacity of an Unknown Metal through Calorimetry** *Calorimetry Problems, Thermochemistry Practice, Specific Heat Capacity, Enthalpy Fusion, Chemistry Specific Heat - Solving for the Final Temperature* *Calorimetry Calculations Specific Heat Solving for Specific Heat of a Substance Heat Capacity and Specific Heat—Chemistry Tutorial* *change in temperature calculations Heat Capacity and Specific Heat | Doc Physics AP Specific Heat (Final Temp. Metal Dropped into Water) Advanced Specific Heat Example Problems* **Specific Heat Example Problems** *Specific Heat Specific Heat Calculations Enthalpy of Formation Reaction \u0026 Heat of Combustion, Enthalpy Change Problems Chemistry* **How to Calculate BTUs for Heating Water** *Latent Heat of Fusion and Vaporization, Specific Heat Capacity \u0026 Calorimetry - Physics Thermodynamics: Calculating Latent and Specific Heat, Example Problem 7.2a Calculating specific heat capacity*

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Calculating the Specific Heat of a Hot Piece of Metal Dropped into Water

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2) Solve for the heat required to change the water into steam (no change in temp). 3) Calculate the heat required to change the temperature of the steam from 100.0 oC to 110.0 oC. 4) To get the heat required for the whole process, ____ the calculated heats from above. $Q = m \times \Delta H_{\text{vap}}$ $Q = 1000. \text{ g} \times 2260 \text{ J/g} = 2,260,000 \text{ J}$ $Q = m \times C \times \Delta t$

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