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Gravimetric Analysis Multiple Choice Questions

A gravimetric analysis is performed to determine the sodium chloride content of a mineral water by precipitating the chloride ions as silver chloride. The experiment is conducted carefully but the result obtained by precipitation differs considerably from the result obtained through atomic absorption spectroscopy.

Review Questions

Stoichiometry, volumetric and gravimetric analyses MULTIPLE CHOICE QUESTIONS Select the one best answer for each question. 1. In the reaction $\text{CaCO}_3 + 2\text{HNO}_3 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{CO}_2 + \text{H}_2\text{O}$, how many milliliters of 0.1015 M HNO_3 are required to react with $0.9857 \text{ grams of CaCO}_3$? (a) 0.1940 mL (b) 97.04 mL (c) 194.1 mL 2.

Chemistry 213 Assignment No. 2 Name: Summer 2010

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Questions And Answers For Gravimetric Analysis

Gravimetric Analysis What is Gravimetric Analysis? Gravimetric analysis is a method in analytical chemistry to determine the quantity of analyte based on the mass of a solid. Example: Measuring the solids suspended in the water sample - Once a known volume of water is filtered, the collected solids are weighed.

Gravimetric Analysis Principle with Types, Advantages and ...

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