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First we determine the mole ratio (remember unknown species on top), H 2 S: H 2 = 8:8 = Number of moles of given species x mole ratio = number of moles required of unknown. So 3.5 x 1 = 3.5 moles of H 2 S could be produced. II. What would be the mass of this maximum yield? To answer this we need to convert our mole value for H 2 S into a mass value.

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moles Cu reacted = 0.1/63.5 = 1.574 x 10-3. 2. mass of silver = 52.1 - 50.1 = 2.00g. moles Ag produced = 2/108 = 8.19 x 10-1. 3. ratio Ag:Cu = 8.19 x 10-1 : 1.574 x 10-3 = 819 x 10-3 : 1.574 x 10-3...

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