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a.) C 5 H 12. b.) C 10 H 22. c.) C 2 H 2. 45. Ethanol (CH 3 CH 2 OH) and propane (C 5 H 12) have similar molar masses. Does that mean they have the same boiling point? Explain. 46. For each of the pairs below, circle the substance with the higher boiling point: a.) I 2 and F 2. b.) HF and LiF . c.) CH 2 O 2 and CCl 4. 47.

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Chapter 12 – Introduction to Organic Chemistry: Hydrocarbons Section 12.1 – Alkanes Goal: Identify properties characteristic of organic or inorganic compounds. Summary: Organic compounds: always contain carbon and hydrogen (thought sometimes other nonmetals as well.) Inorganic compounds: all other compounds. Often ionic (metal + nonmetal)

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(12) disulfide bridges; salt bridges; H-bonding; dispersion forces; (metal-ion coordination is the 5th) (13) Cysteine (Cys) is the amino acid that can form disulfide bridges. (14) The acidic amino acids, aspartic acid (Asp) & glutamic acid (Glu) and the basic amino acids, lysine (Lys), arginine (Arg), and histidine (His) can form salt bridges because their side chains are ionized.