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### Chapter 11 Gases - An Introduction to Chemistry

Internet: Gas Stoichiometry Shortcut Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

### Chapter 11 - Gases

SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure — orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c) ripled. 7-0 (d) unchanged. Rank the following pressures in increasing order. (c) 76 torr (a) 50 kPa O, OOictbv-x

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Relating to ESRD 20 - Renal Dialysis Items and Services 20.1 - Composite Rate Items and Services 20.2 - Laboratory Services 20.3 - Drugs and Biologicals 20.3.1 - Drug Designation Process 20.4 - Equipment and Supplies

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