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Sodium sulfate is a typical electrostatically bonded ionic sulfate. The existence of free sulfate ions in solution is indicated by the easy formation of insoluble sulfates when these solutions are treated with Ba^{2+} or

Pb 2+ salts: $\text{Na}_2\text{SO}_4 + \text{BaCl}_2 \rightarrow 2\text{NaCl} + \text{BaSO}_4$. Sodium sulfate is unreactive toward most oxidizing or reducing agents.

Sodium sulfate - Wikipedia

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