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What Is The Kinetic Theory

Definition of kinetic theory. : either of two theories in physics based on the fact that the minute particles of a substance are in vigorous motion: a : a theory that the temperature of a substance increases with an increase in either the average kinetic energy of the particles or the average potential energy of separation (as in fusion) of the particles or in both when heat is added.

Kinetic Theory | Definition of Kinetic Theory by Merriam ...

kinetic theory A fundamental theory of matter that explains physical properties in terms of the motion of atoms and molecules. In kinetic theory, properties such as pressure and temperature are viewed as statistical properties of the overall behavior of large numbers of particles.

Kinetic theory | Definition of Kinetic theory at ...

kinetic theory. n. A theory of the thermodynamic behavior of matter, especially the relationships among pressure, volume, and temperature in gases, based on the dependence of temperature on the kinetic energy of the rapidly moving particles of a substance. The theory uses statistical mechanics under the presumption that energy and momentum are conserved in all collisions between particles.

Kinetic theory - definition of kinetic theory by The Free ...

Kinetic theory of gases, a theory based on a simplified molecular or particle description of a gas, from which many gross properties of the gas can be derived. Such a model describes a perfect gas and its properties and is a reasonable approximation to a real gas.

kinetic theory of gases | Definition, Assumptions, & Facts ...

Kinetic theory is a scientific theory regarding the nature of gases. The theory goes by many names, including the kinetic theory of gases, kinetic-molecular theory, collision theory, and the kinetic-molecular theory of gases.

What is Kinetic Theory? (with pictures) - wiseGEEK

The kinetic theory of matter states that all matter is made of small particles that are in random motion and that have space between them. This means that no matter what phase matter is in, it is...

The Kinetic Theory of Matter: Definition & The Four States ...

The kinetic theory of gases is a historically significant, but simple, model of the thermodynamic behavior of gases, with which many principal concepts of thermodynamics were established. The model describes a gas as a large number of identical submicroscopic particles, all of which are in constant, rapid, random motion. Their size is assumed to be much smaller than the average distance between the particles. The particles undergo random elastic collisions between themselves and with the enclosi

Kinetic theory of gases - Wikipedia

Edexcel Spec Yr10: kinetic theory . Save for later. What is Kinetic Theory

What is Kinetic Theory | Teaching Resources

The kinetic theory is used description of a gas, from which many properties of gases can be derived. Using the kinetic theory the independent motion of molecules of gases to their pressure, volume ...

What is kinetic theory? - Answers

The kinetic particle theory explains the properties of the different states of matter. The particles in solids, liquids and gases have different amounts of energy. They are arranged differently and...

Kinetic particle theory - Kinetic particle theory and ...

Kinetic theory or kinetic theory of gases attempts to explain overall properties of gases, such as pressure, temperature, or volume, by considering their molecular composition and motion. The theory basically states that pressure is not caused by molecules pushing each other away, like earlier scientists thought.

Kinetic theory - Simple English Wikipedia, the free ...

Kinetic Molecular Theory states that gas particles are in constant motion and exhibit perfectly elastic collisions. Kinetic Molecular Theory can be used to explain both Charles’ and Boyle’s Laws. The average kinetic energy of a collection of gas particles is directly proportional to absolute temperature only.

Kinetic Molecular Theory | Boundless Chemistry

The Kinetic Theory: A Microscopic Description of Matter The kinetic molecular theory of matter offers a description of the microscopic properties of atoms (or molecules) and their interactions, leading to observable macroscopic properties (such as pressure, volume, temperature).

The Kinetic Molecular Theory of Matter | Introduction to ...

Kinetic theory Learn about kinetic theory, which includes using the Celsius and Kelvin scales, the relationship between pressure, temperature and volume in gas, and energy changes when changes in...

Pressure - Kinetic theory - WJEC - GCSE Physics (Single ...

noun Physics. a theory that the temperature of a body is determined by the average kinetic energy of its particles and that an inflow of heat increases this energy.

Kinetic theory of heat | Definition of Kinetic theory of ...

In industrial glass: Kinetic arguments ...on structural criteria but on kinetic theories, which are based on the nucleation and crystal-growth factors outlined in the section Volume and temperature changes. After considering these factors, the glassmaker generates a time-temperature-transformation (T-T-T) diagram.

Kinetic theory | physics | Britannica

Kinetic theory or kinetic theory of gases attempts to explain overall properties of gases, such as pressure, temperature, or volume, by considering their molecular composition and motion. The theory basically states that pressure is not caused by molecules pushing each other away, like earlier scientists thought.

Kinetic theory Facts for Kids | KidzSearch.com

Kinetic theory of gases, an account of gas properties in terms of motion and interaction of submicroscopic particles in gases. Phonon, explaining properties of solids in terms of quantal collection and interactions of submicroscopic particles Free electron model, a model for the behavior of charge carriers in a metallic solid