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chemical substances, and which play an important part due to the consequences they have on the environment : fires and explosions, tropospheric pollution, greenhouse effect, hole in the stratospheric ozone layer.

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The thermal decomposition reaction of 1,2-dibromopropane (12DBP) has been studied at temperatures from 592.2 to 636.2 K over the pressure range 0.23–24.5 Torr. The main products were four different bromopropenes and hydrogen bromide. Propylene as a minor product was also observed by the concurrent heterogeneous deb

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Gas-phase reactions of bis(.eta.5-cyclopentadienyl)methylzirconium¹⁺ with dihydrogen, ethylene, and propylene. Journal of the American Chemical Society 1988 , 110 (12) , 4038-4039.

Gas-phase electron transfer: thermal self-exchange and ...

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Gas-phase reactions can be classified as intramolecular or intermolecular reactions. Intramolecular reactions involve decomposition of precursor molecules into activated species, which are later involved in the CVD process. Hydrogen elimination is an example of such a process: $(\text{CH}_3)_3\text{CAsH}_2(\text{g}) \rightarrow (\text{CH}_3)_2\text{C}=\text{CH}_2(\text{g}) + \text{AsH}_3(\text{g})$.

Gas Phase Reaction - an overview | ScienceDirect Topics

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Ethylene oxide, called oxirane by IUPAC, is an organic compound with the formula C_2H_4O . It is a cyclic ether and the simplest epoxide: a three-membered ring consisting of one oxygen atom and two carbon atoms. Ethylene oxide is a colorless and flammable gas with a faintly sweet odor. Because it is a strained ring, ethylene oxide easily participates in a number of addition reactions that ...

Ethylene oxide - Wikipedia

Chemical Engineering Q&A Library In the gas-phase reaction $2A + B \rightleftharpoons 3C + 2D$, it was found that when 1.50 mole A, 2.00 mole B and 1.00 mole D were mixed and allowed to come to equilibrium at 25°C, the resulting mixture contained 1.20 mol C at a total pressure of 2.00 bar. Calculate (a) the mole fractions of each species at equilibrium x_A , x_B , x_C , and x_D .

Answered: In the gas-phase reaction $2A + B \rightleftharpoons 3C + \dots$ | bartleby

Investigations of gas-phase proton transfer reactions have been performed on protein molecular ions generated by electrospray ionization (ESI). Their reactions were studied in a heated capillary inlet/reactor prior to expansion into a quadrupole mass spectrometer. Results from investigations involving protonated horse heart cytochrome c and H, O suggest that Coulombic effects can lower ...

Gas-phase proton transfer reactions involving multiply ...

The kinetics of the thermal reactions of 1-methylbicyclo[3.2.0]hept-2-en-7-one have been studied in the gas phase over the temperature range 489–565 K. The primary reaction appears to occur via two parallel first-order homogeneous isomerizations to yield 5-methylbicyclo[2.2.1]hept-5-en-2-one and 2-methylhepta-1,3,6-trien-1-one.

The gas-phase thermal reactions of 1-methylbicyclo[3.2.0

...

The hydrogenation of carbon dioxide involves the activation of the thermodynamically very stable molecule CO_2 and formation of a C-H bond. Herein, we report that HCO_2^- and CO can be formed in the thermal reaction of CO_2 with a diatomic metal hydride species, FeH. The FeH-anions were produced by laser ablation, and the reaction with CO_2 was analyzed by mass spectrometry and quantum-chemical ...

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Study of the Thermal Decarboxylation of α,α -Difluoro β -Lactones.
January 1997 · The Journal of Organic Chemistry.

14. Gas phase reactions for chemical CO₂ upgrading ...

Gas-phase elimination reactions of pentane-2,4-dione, methyl acetoacetate, ethyl acetoacetate, 3-phenylhydroazopentane-2,4-dione, and ethyl 3-oxo-2-phenyl-hydrazonebutyrate have been measured in the ...