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ChemicalEquations Lesson Objectives • Describe chemical reactions using word equations. • Know the correct symbols to use in order to write skeleton equations for chemical reactions. • Use coefficients to balance chemical equations so that the law of conservation of mass is followed.

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Section 11.1 –Describing Chemical Reactions. The reactants are written on the left and the products on the right. The arrow that separates them is called yield. Reactants Products. Symbols in Equations. Symbol Meaning " yields D reversible reaction (s) solid (l) liquid (g) gas (aq) aqueous " catalyst " heat. Pt .

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KSEEB Class 8 Science Chapter 11 Additional Questions & Answers. Question 1. Give an example of a complex chemical reaction. Answer: 'Photosynthesis' is a complex chemical reaction. Question 2. Give an example of simple chemical reactions. Answer: Rusting of iron and combustion of fuels are simply chemical reactions. Question 3.

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Chapter 11 Chemical reactions. chemical equation. skeleton equation. catalyst. coefficient. A short, easy way to show a chemical reaction, using symbols. a chemical equation that does not indicate the relative amount.... a substance that initiates or accelerates a chemical reaction....

chemical reactions chapter 11 Flashcards and Study Sets

... To describe a chemical reaction. To calculate the quantities of compounds produced or consumed in a chemical reaction. As

shown in Figure 11.3.1, applying a small amount of heat to a pile of orange ammonium dichromate powder results in a vigorous reaction known as the ammonium dichromate volcano.

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balanced equation. A _____ is an equation where each side of the equation has the same number of atoms of each element, and mass is conserved. word equation. Iron + oxygen --> iron (III) oxide is an example of a(n) _____. chemical equation. $\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_2\text{O}_3$ is an example of a(n) _____.

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reaction. combustion reaction. a reaction in which oxygen reacts with another substance, often producing light or heat. reactant.

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