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**How Small Is An
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light

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Mixtures!
(Classification of
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from an atom?

Photons, light \u0026amp; electromagnetic waves
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Atomic Radius,
Ionization Energy, and
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Electron Energy and Light

Electron Energy and Light 1... Flinn Scientific, Inc. and HSPIThe POGIL Project.. (electricity) when they move from

lower energy levels
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reveal the behavior of
electrons in an atom?
Type your answers in
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before completing the

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1. Select the "Color &
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choose highlighter,
choose the appropriate
color, and trace the
arrows in Model 1 and
shade in the table with
the appropriate color.

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Light 3 11. Consider
the hydrogen spectrum
in Model 2.

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Which of the electron
transitions involves the
most energy? $n=6$ to
 $n=2$, violet light, 410
nm. b. Explain why this
transition involves the
most energy based on
your understanding of
the attractive forces
between the electrons
and protons in the
atom. As the electron
moves toward the
nucleus of the atom

there is a stronger
attraction.

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PLAY. Match. Gravity.
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What happens when
white light passes
through a prism. The
light bends as it passes
through the prism and
then the white light
separates into different
colors with different
wavelengths.

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sentences describes
the relationship
between wavelength
and energy of light?
(#6c) answer choices .
As wavelength gets
shorter the energy ...

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