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Combustion - a fuel reacts with oxygen
to release carbon dioxide, water, and
heat. $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ Single replacement - an
element replaces another element in a
compound. $\text{KCl} + \text{F}_2 \rightarrow \text{KF} + \text{Cl}_2$ $\text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$, $\text{Cu} + \text{AgNO}_3 \rightarrow \text{Cu}(\text{NO}_3)_2 + \text{Ag}$ Double replacement -
two compounds switch parts with one
another. $\text{AgNO}_3 + \text{K}_2\text{SO}_4 \rightarrow \text{Ag}_2\text{SO}_4 + \text{KNO}_3$ $\text{Mg}(\text{OH})_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$...

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Vocabulary: coefficient, combustion, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript, synthesis Prior Knowledge Questions (Do these BEFORE using the Gizmo.) The scouts are making s'mores out of toasted marshmallows, chocolate, and ...

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2 Gizmo Warm-up In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$. How many atoms are in a hydrogen molecule (H_2)? O. 2

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Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction: $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O}$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 + 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? 3 4.

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Student exploration balancing chemical equations gizmo ...

Student Exploration: Balancing Chemical Equations. DEFINE Vocabulary: coefficient. Combination- a chemical reaction when two reactions combine to make one. Compound- a substance consisting of two or more elements . Decomposition- the separation of a substance into a simpler form . Double replacement- multiple elements that are getting swapped

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