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Types of Chemical Reactions Answers Balance each of the following reactions and identify each type of reaction: 1. 2 NaBr + Ca(OH) 2 CaBr 2 + 2 NaOH double displacement 2. 2 NH 3 + H 2 SO 4 (NH 4) 2 SO 4 synthesis 3. 4 C 5 H 9 O + 29 O 2 20 CO 2 + 18 H 2 O combustion 4. 3 Pb + 2 H 3 PO 4 3 H 2 + Pb 3 (PO 4) 2 single displacement 5. Li 3 N + 3 NH 4 NO 3 3 LiNO 3 + (NH 4) 3

Types of Chemical Reaction Worksheet

Types of Reactions: Identify the type of chemical reaction shown.(pg. 356-367) 25.CH 4 + 2O 2 2H 2 O + CO 2 Type: __combustion__ 26.LiCl + NaF LiF + NaCl Type: ____ 27.Ba(OH) 2 + 2Na 2NaOH + Ba Type: Cationic Single R. 28.CaCO 3 CaO + CO 2 Type: ____ 29.MgO + H 2 O Mg(OH) 2 Type: __Synthesis__

Unit 5: Chemical Reactions Homework Packet (85 points)

I CAN balance a chemical equation 5. I CAN identify and describe the six types of chemical reactions based on their reactants and products 6. Given a chemical reaction, I CAN classify the type based on the reactants and products 7. I CAN describe the processes of oxidation and reduction; I CAN identify which elements are oxidized and reduced. 8.

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Different Types of Chemical Reactions. The 5 primary types of chemical reactions are: Combination reaction; Decomposition reaction; Displacement reaction; Double Displacement reaction; Precipitation Reaction; 1. Combination Reaction. A reaction in which two or more reactants combine to form a single product is known as a combination reaction.

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The main four types of reactions are direct combination, analysis reaction, single displacement, and double displacement. If you're asked the five main types of reactions, it is these four and then either acid-base or redox (depending who you ask). Keep in mind, a specific chemical reaction may fall into more than one category.

Types of Chemical Reactions (With Examples)

pearson education chemical reactions packet Chemical Reactions Name: ____ Block: ____ Learning Targets 1. I CAN state and describe the Law of Conservation of Matter as it pertains to chemical reactions 2. I CAN explain the significance of the components of a chemical reaction (reactants, products, arrow, states of matter, etc). 3.

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Major Types of Chemical Reactions Five major types of chemical reactions: synthesis, decomposition, combustion, single replacement (also called single displacement) and double replacement (also called double displacement). In a synthesis reaction, a compound is made from more simple materials. In a decomposition reaction, a compound breaks down into simpler elements or compounds.

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