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Define Solution Chemistry Terms

A solution is a homogeneous mixture of two or more substances. A solution may exist in any phase . A solution consists of a solute and a solvent. The solute is the substance that is dissolved in the solvent. The amount of solute that can be dissolved in solvent is called its solubility.

Solution Definition in Chemistry - ThoughtCo

Solution, in chemistry, a homogenous mixture of two or more substances in relative amounts that can be varied continuously up to what is called the limit of solubility. The term solution is commonly applied to the liquid state of matter, but solutions of gases and solids are possible.

solution | Definition & Examples | Britannica

A solution is a homogeneous mixture of two substances—that is, it has the same distribution of particles throughout. Technically speaking, a solution consists of a mixture of one or more solutes dissolved in a solvent. The particles of solute and solvent are molecules or ions, with one or more solvent molecules bound to each solute particle.

Solution (chemistry) - definition of Solution (chemistry ...

A solution in chemistry is a special type of homogeneous mixture composed of two or more substances. A liquid is a soluble material in another substance, known as a solvent, in such a mixture. To learn more about the topic, register with BYJU'S and download BYJU'S - The Learning App.

Solution - Definition, Properties, Types, Videos & Examples

Search the Dictionary for More Terms. Return to top of page. Copyright © 2020 chemicool.com Chemistry Dictionary | Birth of the Elements | Tools | Periodic Table ...

Definition of solution - Chemistry Dictionary

A standard solution is any chemical solution which has a precisely known concentration. Similarly, a solution of known concentration has been standardized. To prepare a standard solution, a known mass of solute is dissolved and the solution is diluted to a precise volume. Standard solution concentration is usually expressed in terms of molarity (M) or moles per liter (mol/L).

Standard Solution Definition - Chemistry Glossary

A solution is a homogeneous mixture of two or more substances. The particles of solute in a solution

cannot be seen by the naked eye. A solution does not allow beams of light to scatter. A solution is stable. The solute from a solution cannot be separated by filtration (or mechanically). It is composed of only one phase. Types of Solute

Solute (Chemistry) - Definition, Examples, Types with Videos

solution the process by which a gas, liquid, or solid is dispersed homogeneously in a gas, liquid, or solid without chemical... such a substance, as dissolved sugar or salt in solution. a homogeneous, molecular mixture of two or more substances.

Solution | Definition of Solution at Dictionary.com

solvent - This is the liquid that dissolves a solute in solution. Technically, you can dissolve gases into liquids or into other gases, too. When making a solution where both substances are in the same phase (e.g., liquid-liquid), the solvent is the largest component of the solution.

Chemistry Vocabulary - Definitions of Chemistry Terms

This glossary of chemistry terms is a list of terms and definitions relevant to chemistry, including chemical laws, diagrams and formulae, laboratory tools, glassware, and equipment.Chemistry is a physical science concerned with the composition, structure, and properties of matter, as well as the changes it undergoes during chemical reactions; it features an extensive vocabulary and a ...

Glossary of chemistry terms - Wikipedia

Solubility is defined as the maximum quantity of a substance that can be dissolved in another. It is the maximum amount of solute that can be dissolved in a solvent at equilibrium, which produces a saturated solution.

Solubility Definition in Chemistry - ThoughtCo

acid any substance that dissociates in water to yield a sour corrosive solution containing hydrogen ions, having a pH of less than 7, and turning litmus red alcohol a colourless flammable liquid, the active principle of intoxicating drinks, produced by the fermentation of sugars, esp glucose, and

used as a solvent and in the manufacture of organic chemicals.

Chemistry: terms used in chemistry Word Lists | Collins ...

In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent.

Solution - Wikipedia

An unsaturated solution is a chemical solution in which the solute concentration is lower than its equilibrium solubility. All of the solute dissolves in the solvent. When a solute (often a solid) is added to a solvent (often a liquid), two processes occur simultaneously. Dissolution is the dissolving of the solute into the solvent.

What Is an Unsaturated Solution in Chemistry?

solution, in chemistry, homogeneous mixture mixture, in chemistry, a physical combination of two or more pure substances (i.e., elements or compounds). A mixture is distinguished from a compound, which is formed by the chemical combination of two or more pure substances in a fixed, definite proportion.

Solution (chemistry) | Article about Solution (chemistry ...

In analytical chemistry, a standard solution is a solution containing a precisely known concentration of an element or a substance. A known weight of solute is dissolved to make a specific volume. It is prepared using a standard substance, such as a primary standard.

Standard solution - Wikipedia

In chemistry, a stock solution is a large volume of common reagent, such as hydrochloric acid or sodium hydroxide, at a standardized concentration. This term is commonly used in analytical chemistry for procedures such as titrations, where it is important that exact concentrations of solutions are used.