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table in order of their atomic number (proton number). Most of the mass of an atom is in the nucleus The total number of protons and neutrons in an atom is called its mass number To work out the number of neutrons in an atom subtract the atomic number from the mass number Example Beryllium : atomic number 4, mass number 9. It has 4 protons, 4 electrons

Revision Guide: 4.1 Atomic Structure and the Periodic Table

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Atomic Structure was the second of the four starter units in the Independent Learning Project for Advanced Chemistry (ILPAC) scheme that was sponsored by the Inner London Education Authority and first published by John Murray in 1983. This was a study guide, not a textbook.

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